Chemwatch: 4661-20	
	Issue Date: 11/12/2018
Version No: 5.1.1.1	Print Date: 06/21/2019
Safety Data Sheet according to OSHA HazCom Standard (2012) requirements	L.GHS.USA.EN

SECTION 1 IDENTIFICATION

Product Identifier

Product name	API Pond Aquatic Plant Food Tablets
Synonyms	Formula 10-12-8
Other means of identification	Not Available

Recommended use of the chemical and restrictions on use

Relevant identified uses	Use according to manufacturer's directions.
	For product 185.

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	Mars Fishcare North America, Inc.
Address	50 E. Hamilton Street, Chalfont PA 18914 United States
Telephone	215 822 8181
Fax	215 997 1290
Website	Not Available
Email	Not Available

Emergency phone number

Association / Organisation	Mars Fishcare North America, Inc.
Emergency telephone numbers	ChemTel: 1-800-255-3924
Other emergency telephone numbers	ChemTel: 1-813-248-0585

SECTION 2 HAZARD(S) IDENTIFICATION

SIGNAL WORD

WARNING

Classification of the substance or mixture

NFPA 704 diamond



Т

Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification	Skin Corrosion/Irritation Category 2, Eye Irritation Category 2A, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Chronic Aquatic Hazard Category 4
Label elements	
Hazard pictogram(s)	

Hazard statement(s)

H315	Causes skin irritation.
H319	Causes serious eye irritation.
H335	May cause respiratory irritation.
H413	May cause long lasting harmful effects to aquatic life.

Hazard(s) not otherwise classified

Not Applicable

Precautionary statement(s) General

P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.
P103	Read label before use.

Precautionary statement(s) Prevention

P271	Use only outdoors or in a well-ventilated area.
P261	Avoid breathing dust/fumes.
P273	Avoid release to the environment.
P280	Wear protective gloves/protective clothing/eye protection/face protection.

Precautionary statement(s) Response

P362	Take off contaminated clothing and wash before reuse.
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P312	Call a POISON CENTER or doctor/physician if you feel unwell.
P337+P313	If eye irritation persists: Get medical advice/attention.
P302+P352	IF ON SKIN: Wash with plenty of soap and water.
P304+P340	IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.
P332+P313	If skin irritation occurs: Get medical advice/attention.

Precautionary statement(s) Storage

P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

Precautionary statement(s) Disposal

P501	Dispose of contents/container in accordance with local regulations.
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SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
1302-78-9	30-60	bentonite
7783-28-0	10-30	diammonium phosphate
Not Available	<20	nitrate, as
7757-79-1		potassium nitrate
517-25-9	1-10	trinitromethane
Not Available	1-5	nutrients, proprietary
557-04-0	<0.5	magnesium stearate
7732-18-5	<1	DI METERED WATER

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

- ▶ There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Special protective equipment and precautions for fire-fighters

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water courses. Use fire fighting procedures suitable for surrounding area. DO NOT approach containers suspected to be hot. Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use.
Fire/Explosion Hazard	 Non combustible. Not considered a significant fire risk, however containers may burn. Decomposition may produce toxic fumes of: carbon dioxide (CO2) nitrogen oxides (NOx) other pyrolysis products typical of burning organic material. May emit poisonous fumes. May emit corrosive fumes.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Clean up all spills immediately. Secure load if safe to do so. Bundle/collect recoverable product. Collect remaining material in containers with covers for disposal.
Major Spills	 Clean up all spills immediately. Wear protective clothing, safety glasses, dust mask, gloves. Secure load if safe to do so. Bundle/collect recoverable product. Use dry clean up procedures and avoid generating dust. Vacuum up (consider explosion-proof machines designed to be grounded during storage and use). Water may be used to prevent dusting. Collect remaining material in containers with covers for disposal. Flush spill area with water.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Safe handling	 Limit all unnecessary personal contact. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. When handling DO NOT eat, drink or smoke. Always wash hands with soap and water after handling. Avoid physical damage to containers. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this SDS.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS.

Conditions for safe storage, including any incompatibilities

Suitable container	 Polyethylene or polypropylene container. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	 Phosphates are incompatible with oxidising and reducing agents. Phosphates are susceptible to formation of highly toxic and flammable phosphine gas in the presence of strong reducing agents such as hydrides. Partial oxidation of phosphates by oxidizing agents may result in the release of toxic phosphorus oxides. Avoid reaction with oxidising agents



X — Must not be stored together

0 — May be stored together with specific preventions

+ — May be stored together

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)						
INGREDIENT DATA						
Source	Ingredient	Material name	TWA	STEL	Peak	Notes

TEEL-3 2,000 mg/m3 2,000 mg/m3 600 mg/m3

API Pond Aquatic Plant Food Tablets

US ACGIH Threshold Limit Values (TLV)	magnesium stearate	* Stearates(J)	10; 3 mg/m3	Not Available	Not Available	TLV® Basis: LRT irr
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EMERGENCY LIMITS					
Ingredient	Material name		TEEL-1	TEEL-2	
bentonite	Montmorillonite		30 mg/m3	330 mg/m3	
diammonium phosphate	Ammonium phosphate dibasic; (Diammonium phosphate)		30 mg/m3	330 mg/m3	
potassium nitrate	Potassium nitrate		9 mg/m3	100 mg/m3	
Ingredient	Original IDLH	Revis	ed IDLH		
bentonite	Not Available	Not Av	vailable		
diammonium phosphate	Not Available	Not Av	vailable		
potassium nitrate	Not Available	Not Av	vailable		
trinitromethane	Not Available	Not Av	vailable		
magnesium stearate	Not Available	Not Av	vailable		
DI METERED WATER	Not Available	Not Av	vailable		

MATERIAL DATA

Exposure controls

	Engineering controls are used to remove a hazard or place a barrier between the engineering controls can be highly effective in protecting workers and will type to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is Enclosure and/or isolation of emission source which keeps a selected hazard ventilation that strategically "adds" and "removes" air in the work environment contaminant if designed properly. The design of a ventilation system must match contaminant in use. Employers may need to use multiple types of controls to prevent employee of General exhaust is adequate under normal operating conditions. Local exhaust circumstances. If risk of overexposure exists, wear approved respirator. Corr protection. Provide adequate ventilation in warehouse or closed storage areas workplace possess varying "escape" velocities which, in turn, determine the "required to effectively remove the contaminant.	ically be independent of done to reduce the risk "physically" away from t. Ventilation can remov atch the particular proce overexposure. st ventilation may be rec rect fit is essential to ob s. Air contaminants gene	worker interactions the worker and e or dilute an air ess and chemical or quired in specific tain adequate erated in the		
	Type of Contaminant:		Air Speed:		
	solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min)			
Appropriate engineering controls	aerosols, fumes from pouring operations, intermittent container filling, low sp transfers, welding, spray drift, plating acid fumes, pickling (released at low v active generation)	0.5-1 m/s (100-200 f/min.)			
	direct spray, spray painting in shallow booths, drum filling, conveyer loading discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)			
	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (relevelocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)			
	Within each range the appropriate value depends on:				
	Lower end of the range	Upper end of the rang	je		
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents			
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of hi	high toxicity		
	3: Intermittent, low production.	3: High production, h	heavy use		
	4: Large hood or large air mass in motion	ontrol only			
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.				

Page 6 of 13

API Pond Aquatic Plant Food Tablets

Personal protection	
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber
Body protection	See Other protection below
Other protection	 Overalls. P.V.C. apron. Barrier cream. Skin cleansing cream. Eye wash unit.

Respiratory protection

Type AX Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Pale mottled bluish-green odourless cylindrical tablet with brown specks; slowly soluble in water for slow release.			
Physical state	Manufactured	Relative density (Water = 1)	Not Applicable	
Odour	Not Available	Partition coefficient n-octanol / water	Not Available	
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable	
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available	
Melting point / freezing point (°C)	Not Applicable	Viscosity (cSt)	Not Applicable	
Initial boiling point and boiling range (°C)	Not Applicable	Molecular weight (g/mol)	Not Applicable	
Flash point (°C)	Not Applicable	Taste	Not Available	
Evaporation rate	Not Applicable	Explosive properties	Not Available	
Flammability	Not Applicable	Oxidising properties	Not Available	
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Applicable	
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Applicable	
Vapour pressure (kPa)	Not Applicable	Gas group	Not Available	
Solubility in water	Miscible	pH as a solution (1%)	Not Applicable	
Vapour density (Air = 1)	Not Applicable	VOC g/L	Not Available	

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	Product is considered stable and hazardous polymerisation will not occur.

Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled	The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless, adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting.
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual.
Skin Contact	The material is not thought to produce adverse health effects or skin irritation following contact (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting. Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Eye	The material may be irritating to the eye, with prolonged contact causing inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.
Chronic	Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

API Pond Aquatic Plant	TOXICITY	IRRITATION
Food Tablets	Not Available	Not Available
	тохісіту	IRRITATION
bentonite	Oral (rat) LD50: >5000 mg/kg ^[2]	Not Available
	тохісіту	IRRITATION
diammonium phosphate	dermal (rat) LD50: >5000 mg/kg ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
	Oral (rat) LD50: >2000 mg/kg ^[1]	Skin: no adverse effect observed (not irritating) ^[1]
	тохісіту	IRRITATION
potassium nitrate	dermal (rat) LD50: >5000 mg/kg ^[1]	Not Available
	Oral (rat) LD50: >2000 mg/kg ^[1]	
	ТОХІСІТҮ	IRRITATION
trinitromethane	Inhalation (mouse) LC50: 0.4 mg/l/2H ^[2]	Not Available
	ТОХІСІТҮ	IRRITATION
magnesium stearate	Oral (rat) LD50: >10000 mg/kg ^[2]	Not Available
DI METERED WATER	тохісіту	IRRITATION
	Oral (rat) LD50: >90000 mg/kg ^[2]	Not Available
Legend:	1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances	

BENTONITE	for bentonite clays: Bentonite (CAS No. 1302-78-9) consists of a group of clays formed by crystallisation of vitreous volcanic ashes that were deposited in water. The expected acute oral toxicity of bentonite in humans is very low (LD50>15 g/kg). However, severe anterior segment inflammation, uveitis and retrocorneal abscess from eye exposure were reported when bentonite had been used as a prophypaste. In a 33 day dietary (2 and 6%) and a 90 day dietary (1, 3 and 5%) studies in chickens, no changes in behaviour, overall state, clinical and biochemical parameters and electrolytic composition of the blood. Repeat dietary administration of
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	bentonite did not affect calcium or phosphorus weakness, and death with marked changes in Bentonite did not cause fibrosis after 1 year e study, where 5 um particles were intratrachea Bentonite clay dust is believed to be responsi Ingestion of bentonite without adequate liquid Hypokalaemia and microcytic iron-deficiency has been reported to cause myositis.	both calcium and phosphorus n exposure of 60 mg dust (<5 um) ally instilled at 5, 15 and 45 mg/r ble for bronchial asthma in work is may result in intestinal obstrue	netabolism.) in a rat study. However, in a second rat rat, dose-related fibrosis was observed. kers at a processing plant in USA.
	The material may be irritating to the eye, with irritants may produce conjunctivitis. The material may produce respiratory tract irr laryngitis, shortness of breath, headache, nau Unlike most organs, the lung can respond to irritant and then repairing the damage (inflam The repair process (which initially developed	ritation. Symptoms of pulmonary usea, and a burning sensation. a chemical insult or a chemical mation of the lungs may be a co to protect mammalian lungs from	agent, by first removing or neutralising the
TRINITROMETHANE	 impairment of gas exchange, the primary function of the lungs. Therefore prolonged exposure to respiratory irritants may cause sustained breathing difficulties. The material may cause skin irritation after prolonged or repeated exposure and may produce a contact dermatitis (nonallergic). This form of dermatitis is often characterised by skin redness (erythema) and swelling epidermis. Histologically there may be intercellular oedema of the spongy layer (spongiosis) and intracellular oedema of the epidermis. Chemical with the aliphatic nitro group (-C-NO2) have been added to a list of DNA-reactive subgroups recognised by the National Toxicological Program (NTP, U.S. Dept Health and Human Services) for possible carcinogenic activity. Inhalation (mouse) LC50: 800 mg/kg/2h (?) Eye: slight * General anaesthesia, ataxia, dyspnea, degenerative changes in the brain and covering, reproductive system tumours recorded. 		
MAGNESIUM STEARATE	Fatty acid salts are of low acute toxicity. Their skin and eye irritation potential is chain length dependent and decreases with increasing chain length - they are poorly absorbed through the skin nor are they skin sensitisers. The available repeated dose toxicity data demonstrate the low toxicity of the fatty acids and their salts. Also, they are not considered to be mutagenic, genotoxic or carcinogenic, and are not reproductive or developmental toxicants. Accidental ingestion of fatty acid salt containing detergent products is not expected to result in any significant adverse health effects. This assessment is based on toxicological data demonstrating the low acute oral toxicity of fatty acid salts and the fact that not a single fatality has been reported in the UK following accidental ingestion of Consumers and Veterinary Medicine, detergent products were not mentioned as dangerous products with a high incidence if poisoning. The estimated total human exposure to fatty acid salts, from the different exposure scenarios for the handling and use of detergent products containing fatty acid salts, showed a margin of exposure (MOE) of 258,620. This extremely large MOE is large enough to be reassuring with regard to the relatively small variability of the hazard data on which it is based. Also, in the UK, the recommended dietary fatty acid intake by the Department of Health is about 100 g of fatty acids per day or 1.7 g (1700 mg) of fatty acids per kilogram body weight per day. This exposure is several orders of magnitude above that resulting from exposure to fatty acid salts in household cleaning products. Based on the available data, the use of fatty acid salts in household detergent and cleaning products does not raise any safety concerns with regard to consumer		
BENTONITE & DIAMMONIUM PHOSPHATE & TRINITROMETHANE & MAGNESIUM STEARATE	Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.		
BENTONITE & DIAMMONIUM PHOSPHATE & DI METERED WATER	No significant acute toxicological data identified in literature search.		
Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	¥	Reproductivity	×

Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	×
Mutagenicity	×	Aspiration Hazard	×
	Legen	d: 🗙 – Data either not availa	ble or does not fill the criteria for classification

Data available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

API Pond Aquatic Plant Food Tablets	Not Available	Not Available	Not Available		Not	Not
			Not Available		Available	Available
	ENDPOINT	TEST DURATION (HR)	SPECIES		VALUE	SOURCI
bentonite	LC50	96	Fish	1	19000mg/L	4
	ENDPOINT	TEST DURATION (HR)	SPECIES	VAL	.UE	SOURC
	LC50	96	Fish	0.0	01-0.32mg/L	2
liammonium phosphate	EC50	48	Crustacea	>1()0mg/L	2
	EC50	72	Algae or other aquatic plants	>1(00mg/L	2
	NOEC	72	Algae or other aquatic plants	100)mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	V	ALUE	SOURC
	LC50	96	Fish	1	-378mg/L	2
potassium nitrate	EC50	48	Crustacea	4	90mg/L	2
	EC50	96	Algae or other aquatic plants	1	181.887mg/L	3
	NOEC	720	Fish	5	8mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VAI	LUE	SOURC
trinitromethane	LC50	96	Fish	16	579.803mg/L	3
	EC50	96	Algae or other aquatic plants	19	2000mg/L	3
	ENDPOINT	TEST DURATION (HR)	SPECIES		VALUE	SOURCE
magnesium stearate	Not Available	Not Available	Not Available		Not Available	Not Availabl
	ENDPOINT	TEST DURATION (HR)	SPECIES	V	ALUE	SOURC
DI METERED WATER	LC50	96	Fish	89	97.520mg/L	3
	EC50	96	Algae or other aquatic plants	8	768.874mg/L	3

Bioconcentration Data 8. Vendor Data

In air ammonia is persistent whilst, in water, it biodegrades rapidly to nitrate, producing a high oxygen demand. Ammonia is strongly adsorbed to soil. Ammonia is non-persistent in water (half-life 2 days) and is moderately toxic to fish under normal temperature and pH conditions. Ammonia is harmful to aquatic life at low concentrations but does not concentrate in the food chain. Ammonium ions may be toxic to fish at 0.3 mg/l Drinking Water Standards:

0.5 mg/l (UK max.)

1.5 mg/l (WHO Levels)

Soil Guidelines: none available.

Air Quality Standards: none available.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
potassium nitrate	LOW	LOW
trinitromethane	HIGH	HIGH
DI METERED WATER	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
potassium nitrate	LOW (LogKOW = 0.209)
trinitromethane	LOW (LogKOW = -1.8492)
DI METERED WATER	LOW (LogKOW = -1.38)

Mobility in soil

Ingredient	Mobility
potassium nitrate	LOW (KOC = 14.3)
trinitromethane	LOW (KOC = 54.17)
DI METERED WATER	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Naste treatment methods	
Product / Packaging disposal	 Recycle wherever possible or consult manufacturer for recycling options. Consult State Land Waste Authority for disposal. Bury or incinerate residue at an approved site. Recycle containers if possible, or dispose of in an authorised landfill.

SECTION 14 TRANSPORT INFORMATION

Labels Required

Marine Pollutant NO

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

BENTONITE(1302-78-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US DOE Temporary Emergency Exposure Limits (TEELs)	US TSCA Chemical Substance Inventory - Interim List of Active
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	Substances
DIAMMONIUM PHOSPHATE(7783-28-0) IS FOUND ON THE FOLLOWING RE	GULATORY LISTS
GESAMP/EHS Composite List - GESAMP Hazard Profiles	US List of Active Substances Exempt from the TSCA Inventory
IMO IBC Code Chapter 17: Summary of minimum requirements	Notifications (Active-Inactive) Rule
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
Bulk	US TSCA Chemical Substance Inventory - Interim List of Active
US DOE Temporary Emergency Exposure Limits (TEELs)	Substances

POTASSIUM NITRATE(7757-79-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations	US Department of Transportation (DOT), Hazardous Material Table
International Maritime Dangerous Goods Requirements (IMDG Code)	US DOE Temporary Emergency Exposure Limits (TEELs)
United Nations Recommendations on the Transport of Dangerous Goods	US EPCRA Section 313 Chemical List
Model Regulations	US Postal Service (USPS) Hazardous Materials Table: Postal Service
US - Massachusetts - Right To Know Listed Chemicals	Mailability Guide
US - Pennsylvania - Hazardous Substance List	US Postal Service (USPS) Numerical Listing of Proper Shipping Names by
US - Rhode Island Hazardous Substance List	Identification (ID) Number
US Department of Homeland Security (DHS) - Chemical Facility	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
Anti-Terrorism Standards (CFATS) - Chemicals of Interest	US TSCA Chemical Substance Inventory - Interim List of Active
	Substances

TRINITROMETHANE(517-25-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations International Air Transport Association (IATA) Dangerous Goods Regulations	US Department of Transportation (DOT), Hazardous Material Table : Goods Forbidden for Transport
- Prohibited List Passenger and Cargo Aircraft	US Postal Service (USPS) Hazardous Materials Table: Postal Service
International Maritime Dangerous Goods Requirements (IMDG Code)	Mailability Guide
United Nations Recommendations on the Transport of Dangerous Goods Model Regulations	US Postal Service (USPS) Numerical Listing of Proper Shipping Names by Identification (ID) Number
US Department of Transportation (DOT), Hazardous Material Table	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

MAGNESIUM STEARATE(557-04-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - California Permissible Exposure Limits for Chemical Contaminants	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Idaho Toxic Air Pollutants Non- Carcinogenic Increments -	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
Occupational Exposure Limits	US TSCA Chemical Substance Inventory - Interim List of Active
US ACGIH Threshold Limit Values (TLV)	Substances

DI METERED WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

IMO IBC Code Chapter 18: List of products to which the Code does not	US TSCA Chemical Substance Inventory - Interim List of Active
apply	Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

Flammable (Gases, Aerosols, Liquids, or Solids)	No
Gas under pressure	No
Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	No
Reproductive toxicity	No
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	No
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	No
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No
Hazards Not Otherwise Classified	No

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

None Reported

State Regulations

US. CALIFORNIA PROPOSITION 65

None Reported

National Inventory Status

National Inventory	Status
Australia - AICS	No (trinitromethane)
Canada - DSL	No (trinitromethane)
Canada - NDSL	No (bentonite; DI METERED WATER; magnesium stearate; potassium nitrate)
China - IECSC	No (trinitromethane)
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	No (bentonite; trinitromethane)
Korea - KECI	No (trinitromethane)
New Zealand - NZIoC	No (trinitromethane)
Philippines - PICCS	No (trinitromethane)
USA - TSCA	Yes
Taiwan - TCSI	No (trinitromethane)
Mexico - INSQ	No (trinitromethane)
Vietnam - NCI	No (trinitromethane)
Russia - ARIPS	No (trinitromethane)
Thailand - TECI	No (trinitromethane)
Legend:	Yes = All CAS declared ingredients are on the inventory No = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Revision Date	11/12/2018
Initial Date	02/13/2006

SDS Version Summary

Version	Issue Date	Sections Updated
4.1.1.1	06/27/2017	Classification
5.1.1.1	11/12/2018	Name

Other information

Ingredients with multiple cas numbers

Name	CAS No
bentonite	1302-78-9, 11004-12-9, 10043-07-9, 115628-71-2, 12198-92-4, 12199-69-8, 135945-01-6, 37320-72-2, 52623-66-2, 850872-77-4, 67479-91-8, 89382-86-5, 90989-60-9, 85049-30-5, 97862-66-3, 84776-12-5, 70131-50-9, 90989-59-6
diammonium phosphate	7783-28-0, 14265-44-2

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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